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Suppose that normal living tuna contains a certain nucleus that has a half-life of 10 years. Once canned the nuclei begin to decay. If we find a can that has half the expected amount of the nucleus, how old is the can?

- A). We can not tell
- B). 1 year
- C). 10 years
- D). 20 years
- E). ¹/₂ year



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After three half-lives, what fraction of the original material is left? A). 1/2 B). $(1/2) \cdot (1/2) \cdot (1/2) = 1/8$ C). $(1/2) \cdot (1/2) = 1/4$ D). $(1/2) \cdot (1/2) \cdot (1/2) \cdot (1/2) \cdot (1/2) \cdot (1/2) = 1/64$ E). $2^3 = 8$





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The equation for fraction remaining is

$$f = \frac{A}{B} = \left(\frac{1}{2}\right)^C$$

Which letter in the above equation is the number of half-lives?A). AB). BC). C



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DATA: Take the half-life of 14-C to be 6000 years. If we find a sample of old bone that has (1/16) the normal amount of 14-C found in living bone, how old is the bone?

- A). 6000 years
- B). 12,000 years
- C). 24,000 years
- D). (1/16) years
- E). (1/6000)⁴ years



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When rock forms it has almost no 40-Ar in it. However, the radioactive isotope 40-K, which does form in rock, decays to 40-Ar with a half-life of 1.3 billion years.

Suppose we find a rock with a ratio of 40-Ar to 40-K of 1:1. How old is the rock?

A)0 years B) 1.3 billion years C) 2.6 billion years

D) 3.9 billion years E) It is not possible to have that ratio