YOUR NAME:_____

1. Consider the Van der Waals equation of state,

$$P = \frac{NT}{V - Nv_0} - a\left(\frac{N}{V}\right)^2.$$

Which terms affect the second virial coefficient?

(a) v_0

(b) a

- (c) neither v_0 nor a
- (d) both v_0 and a
- 2. At the critical point in the liquid-gas phase transition denote each quantity as being zero/finite/infinite.
 - (a) $dP/dV|_T = (\text{zero} / \text{finite value} / \text{infinite value})$
 - (b) $d^2 P/dV^2|_T = (\text{zero} / \text{finite value} / \text{infinite value})$
 - (c) speed of sound = (zero / finite value / infinite value)
 - (d) the correlation length $\xi = (\text{zero} / \text{finite value} / \text{infinite value})$

3. A diatomic molecule has mass m and moment of inertia I. A dilute gas of such molecules is kept at temperature T and chemical potential μ . Assume the temperature is low enough to ignore any vibrational states. For the following questions, do not assume the temperature is so high that rotational states can be treated as a continuum or so low that the excited states can be ignored. In terms of μ , $\beta = 1/T$, m and I,

- (a) Calculate the density ρ .
- (b) Calculate the pressure P.
- (c) Calculate the energy density E/V.
- (d) Calculate the entropy density S/V.